

The first of these is the fact that the
 C_{60} molecule is a truncated icosahedron,
 which is a polyhedron with 32 faces,
 12 of which are pentagons and 20 are hexagons.
 This structure is very similar to a soccer ball,
 and it is this structure that gives the
 C_{60} molecule its unique properties.

The first of these is the fact that the
 C_{60} molecule is a truncated icosahedron,
 which is a polyhedron with 32 faces, 60
 vertices, and 90 edges. The second is
 the fact that the C_{60} molecule is a
 highly symmetric molecule, with a
 symmetry group of I_h . The third is
 the fact that the C_{60} molecule is a
 highly stable molecule, with a
 dissociation energy of approximately
 10 eV per molecule. The fourth is
 the fact that the C_{60} molecule is a
 highly soluble molecule, with a
 solubility of approximately 1 mg/ml
 in toluene. The fifth is the fact that
 the C_{60} molecule is a highly
 reactive molecule, with a
 reactivity of approximately 10
 eV per molecule. The sixth is the
 fact that the C_{60} molecule is a
 highly conductive molecule, with a
 conductivity of approximately 10
 S/cm. The seventh is the fact that
 the C_{60} molecule is a highly
 magnetic molecule, with a
 magnetic moment of approximately
 10 μ_B per molecule. The eighth is
 the fact that the C_{60} molecule is a
 highly catalytic molecule, with a
 catalytic activity of approximately
 10 $\mu\text{mol/g h}$ per molecule. The
 ninth is the fact that the C_{60}
 molecule is a highly photovoltaic
 molecule, with a photovoltaic
 efficiency of approximately 10% per
 molecule. The tenth is the fact that
 the C_{60} molecule is a highly
 thermoelectric molecule, with a
 thermoelectric coefficient of
 approximately 10 $\mu\text{V/K}$ per molecule.